

Two Triterpenoid Dimers from *Rubus pungens* Camb var. *oldhamii*

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Abstract: Two new triterpenoid saponin dimers, named rubupungenoside A (**1**) and B (**2**), have been isolated from the aerial parts of *Rubus pungens* Camb. var. *oldhamii*. Their structures have been established on the basis of spectroscopic methods and chemical transformations.

Keywords: *Rubus pungens* Camb. var. *oldhamii*, ursene, triterpenoid, dimer.

In the course of our continuous chemical studies of *Rubus* species¹⁻⁴, we have isolated two new triterpenoid dimers (**1**) and (**2**), in their methyl ester forms (**1a**) and (**2a**), from the aerial parts of *Rubus pungens* Camb var. *oldhamii*.

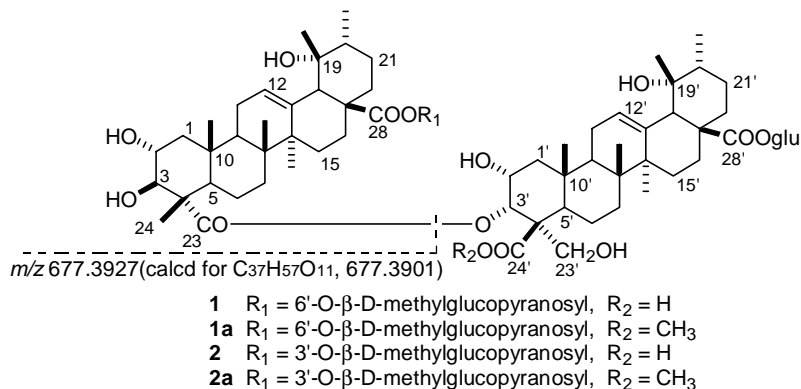
Compound **1a**, amorphous powder, m.p. 231-233°C, $[\alpha]_D^{21} + 12.32$ (*c* 0.35, CH₃OH), gave a positive coloration in the Liebermann-Burchard and Molish tests which suggests that **1a** was a triterpenoid glycoside. The IR spectrum revealed the presence of hydroxyl (3435 cm⁻¹), ester carbonyl (1727 cm⁻¹), and double bond (1646 cm⁻¹) in the molecule. Its molecular formula was determined by negative HRFABMS as C₇₄H₁₁₄O₂₄ (1385.7638, [M-1]⁻, calcd for C₇₄H₁₁₃O₂₄, 1385.7622). Most signals observed in the ¹³C NMR spectrum as doublets (**Table 1**) implied that **1a** might be a triterpenoid dimer.

Hydrolysis of **1a** with 3% NaOH yielded two known triterpenoids **3** and **4**. This suggests that **1a** is a dimer related to **3** and **4**. By the detailed analysis of NMR and EIMS data, **3** and **4** were identified as 2 α , 3 β , 19 α -trihydroxyurs-12-en-23, 28-dioic acid^{2, 5} and 24-methyl ester of 2 α , 3 α , 19 α , 23-tetrahydroxyurs-12-en-24, 28-dioic acid⁶, respectively. Treatment of **3** and **4** with diazomethane yielded dimethyl ester **3a** and **4a**, which were acetylated with Ac₂O / pyridine afforded di-acetylated product **3b** and tri-acetylated product **4b**, respectively. These chemical transformations further confirmed structures of **3** and **4**.

In the ¹H NMR spectrum, H-3 signal appeared at δ 3.40 in **4**, and this signal shifted 1.73 ppm to the lower field at δ 5.13 in **1a**⁷, which implied that 3-OH in **4** is one of the linkage positions. The carbon signal at δ 70.3 (d, C-3') in **4** shifted 3.6 ppm to the lower field at δ 73.9 in **1a**, also indicated that 3-OH in **4** is a linkage position to **1a**. Similarly, the C-23 carbon signal in **3** appeared as a carboxylic carbonyl group at δ 181.0 ppm, while in **1a** this signal appeared as an ester carbonyl group at δ 177.6 in the ¹³C NMR spectrum, revealed that C-23 carboxylic group in **3** is another linkage position for **1a**. Its

HMBC spectrum exhibited cross peaks between C-24' and H-3' (δ 5.13) and 24'-COOCH₃ (δ 3.75), and between C-23 (δ 177.6) and H-3 (δ 4.38), H-24 (δ 1.47), and H-3' (δ 5.13) which further supported the above deduction.

A careful comparison of the ¹H and ¹³C NMR data of **1a** with those of coreanoside F1, a triterpenoid dimer isolated from *Rubus coreanus*⁶, showed that the structures of these two compounds were very similar, except that **1a** had two additional methoxy groups, one was the C-24' ester methyl at δ 51.5 (q), and another was at δ 59.1 (q), connected at C-6 position of the glucosyl moiety, which caused the C-6 signal of glucosyl moiety shifted from *ca* δ 62 ppm in glucosyl to δ 72.5 ppm in 6-methyl-glucosyl of **1a**. Meanwhile, the seven carbon signals of 6-methylglucosyl moiety observed in the ¹³C NMR spectrum of **1a** appeared almost the same as those of 6-O- β -D-methylglucopyranose⁸. These spectral evidences indicated that one of the glucosyls in coreanoside F1 was substituted by a 6-O-methylglucosyl moiety in **1a**. The fragment ion peaks at *m/z* 677.3927 (C₃₇H₅₇O₁₁) and 501.3252 [677 - methylglucosyl]⁻ suggest that 6-O- β -D-methylglucopyranosyl moiety must be attached to C-28 position, and then β -D-glucopyranosyl moiety must be attached to C-28' position. The anomeric proton signals at δ 6.25 (1H, d, *J* = 8.2 Hz) and δ 6.18 (1H, d, *J* = 8.2 Hz) in the ¹H NMR spectrum of **1a** indicated the β -configuration for both of the glucosyl and methylglucosyl moieties.



From all above-mentioned, the structure of **1a** was assigned as shown. The naturally occurring compound should be as **1**, named rubupungenoside A. Its NMR spectral data were assigned by detailed analysis of 2D NMR spectral data (¹H-¹H cosy, HMQC, HMBC, HMQC-TOCSY).

Compound **2a**, another amorphous powder, m.p. 216-218 °C, [α]_D²¹ +15.81 (*c* 0.25, CH₃OH), also gave a positive coloration in the Liebermann-Burchard and Molish tests suggesting that **2a** was another triterpenoid glycoside. The presence of hydroxyl, ester carbonyl and double bond in the molecule was suggested by the absorption bands at 3440, 1727, and 1648 cm⁻¹ in the IR spectrum. Its negative FABMS showed the same molecular and fragment ion peaks as those of **1a** indicated that **2a** has the same molecular formula as **1a** (C₇₄H₁₁₄O₂). This suggestion was confirmed by its very closed NMR data as **1a** (**Table 1**). In the ¹³C NMR spectrum, the differences of **2a** to those of

1a were observed with changes in chemical shifts for C-3 and C-6 positions of C-28 methylglucosyl moiety. The C-6 signal of 28-methylglucosyl in **2a** resonated at δ 62.2 (t), 10.3 ppm upfield compared to that of **1a**, indicated the de-etherification at C-6 position of methylglucosyl group. Furthermore, the signal at δ 88.8 (d) corresponding to the C-3 carbon of 28-methylglucosyl residue in **2a**, which shifted 11.0 ppm to lower field compared to that of **1a**, implied the etherification of C-3 hydroxyl group of 28-methylglucosyl moiety in **2a**. The other carbons of **2a** resonated in almost the same positions as in **1a** (Table 1).

From the above deduction, the structure of **2a** was established as shown. The related natural compound should be as **2**, which was named as rubupungenoside B.

Table 1. ^{13}C NMR (DEPT) data of compounds **1a**, **2a**, **3** and **4**

C no.	1a ^a	2a ^a	3 ^b	4 ^b
1	48.7 (t)	42.8 (t)	48.7 (t)	43.0 (t)
2	69.5 (d)	65.8 (d)	69.9 (d)	66.1 (d)
3	80.8 (d)	73.9 (d)	81.0 (d)	73.4 (d)
4	55.5 (s)	55.5 (s)	55.5 (s)	55.5 (s)
5	52.2 (d)	54.4 (d)	52.4 (d)	54.5 (d)
6	21.2 (t)	20.7 (t)	21.4 (t)	20.7 (t)
7	32.8 (t)	33.3 (t)	33.1 (t)	33.6 (t)
8	40.4 (s)	40.4 (s)	40.7 (s)	40.4 (s)
9	48.5 (d)	48.5 (d)	48.9 (d)	47.3 (d)
10	38.7 (s)	38.3 (s)	39.0 (s)	38.6 (s)
11	24.1 (t)	24.2 (t)	24.3 (t)	24.2 (t)
12	128.2 (d)	127.8 (d)	128.4 (d)	128.1 (d)
13	139.2 (s)	139.1 (s)	139.4 (s)	139.3 (s)
14	42.8 (s)	40.4 (s)	42.3 (s)	40.7 (s)
15	29.0 (t)	29.0 (t)	29.1 (t)	29.3 (t)
16	25.9 (t)	26.0 (t)	26.2 (t)	26.2 (t)
17	48.1 (s)	48.1 (s)	48.9 (s)	48.9 (s)
18	54.3 (d)	54.3 (d)	54.6 (d)	54.5 (d)
19	72.4 (s)	72.4 (s)	72.8 (s)	72.8 (s)
20	42.0 (d)	42.0 (d)	42.3 (d)	42.2 (d)
21	26.8 (t)	26.6 (t)	26.8 (t)	26.8 (t)
22	37.5 (t)	37.5 (t)	37.7 (t)	37.7 (t)
23	177.6 (s)	68.7 (t)	177.6 (s)	68.9 (t)
24	12.8 (q)	175.1 (s)	12.9 (q)	175.3 (s)
25	17.3 (q)	17.3 (q)	17.5 (q)	17.5 (q)
26	16.5 (q)	16.5 (q)	17.5 (q)	16.8 (q)
27	24.4 (q)	24.1 (q)	24.7 (q)	24.6 (q)
28	176.7 (s)	176.7 (s)	177.1 (s)	177.1 (s)
29	26.8 (q)	26.8 (q)	27.1 (q)	27.1 (q)
30	16.5 (q)	14.4 (q)	16.8 (q)	14.6 (q)
24-CO ₂				
Me		51.5 (q)		51.6 (q)
Sugar moiety				
1	95.6 (d)	95.7 (d)	95.9 (d)	95.7 (d)
2	73.8 (d)	73.8 (d)	74.1 (d)	74.1 (d)
3	77.8 (d)	78.3 (d)	88.8 (d)	79.3 (d)
4	71.0 (d)	71.1 (d)	70.9 (d)	71.4 (d)
5	78.7 (d)	79.2 (d)	79.0 (d)	79.0 (d)
6	72.5 (t)	62.2 (t)	62.2 (t)	62.5 (t)
6-OMe	59.1 (q)			

3-OMe	60.9 (q)
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^a recorded in C₅D₅N. ^b recorded in CD₃OD.

Since diazomethane has been used to esterify the carboxylic group because of its difficulty of separation, and methanol has been used as elution in the procedures of purification, **1** and **2** are perhaps artifacts. To confirm whether they are natural products or artifacts, we selected a natural compound 2 α , 3 β , 19 α - trihydroxyurs - 12 - en - 24, 28 - dioic acid - 28 - O - β - D - glucopyranosyl ester (trachelosperoside A-1, previously isolated from *Trachelospermum asiaticum*⁹ and *Rubus pileatus*⁴) as a control test. This compound was treated with diazomethane to afford its 24-methyl ester derivative which was further subjected to CC on silica gel and eluting repeatedly with the same solvents as used in the separation procedures (CHCl₃ : CH₃OH : H₂O, 20 : 5 : 1). The eluted compound was collected and its NMR and MS spectral data revealed that the glucosyl moiety remains unchanged, supporting that **1** and **2** are natural products.

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7. ¹H NMR data (pyridine-*d*₅, 400 MHz): **1a** δ 6.25 (1H, d, *J* = 8.2 Hz, anomeric proton of methylglucosyl moiety), 6.18 (1H, d, *J* = 8.2 Hz, anomeric proton of glucosyl moiety), 5.13 (1H, br.s, H-3'), 3.75 (3H, s, 24'-COOCH₃), 3.50 (3H, s, methyl proton of methylglucosyl group), 1.47 (3H, s, H-24), 1.33 (3H, d, *J* = 6.8 Hz, H-30 or H-30'), 1.17 (3H, d, *J* = 8.3 Hz, H-30' or H-30), 1.60, 1.55, 1.07, 1.05, 1.03, 1.01, 0.99, 0.97 (each 3H, s, H-25, 26, 27, 29, 25', 26', 27', and 29'); **2a**: δ 6.28 (1H, d, *J* = 8.0 Hz, anomeric proton of methylglucosyl group), 6.23 (1H, d, *J* = 8.2 Hz, anomeric proton of glucosyl moiety), 5.21 (1H, br. s, H-3'), 4.88 (2H, d, *J* = 9.5 Hz, H-23'), 4.34 (1H, d, *J* = 8.9 Hz, H-3), 3.89 (3H, s, 24'-COOCH₃), 3.75 (3H, s, methyl proton of methylglucosyl group), 1.37 (3H, d, *J* = 7.1 Hz, H-30 or H-30'), 1.05 (3H, d, *J* = 6.0 Hz, H-30' or H-30), 1.68, 1.64, 1.63, 1.50, 1.19, 1.11, 1.09, 0.99, 0.98 (each 3H, s, H-24, 25, 26, 27, 29, 25', 26', 27', 29).
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